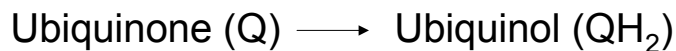


## Chapter 12 - The Citric Acid Cycle

- The citric acid cycle (tricarboxylic acid cycle) is **amphibolic** (both catabolic and anabolic)
- The cycle is involved in the aerobic catabolism of carbohydrates, lipids and amino acids
- Intermediates of the cycle are starting points for many biosynthetic reactions
- Enzymes of the cycle are in the mitochondria (eukaryotes) or the cytosol of bacteria

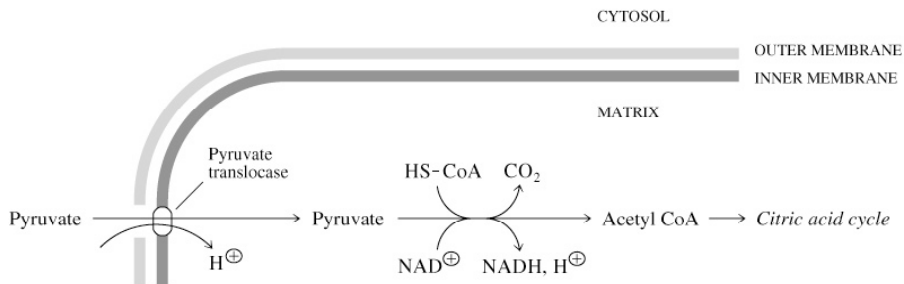
## Energy in the citric acid cycle

- Energy of the oxidation reactions is largely conserved as reducing power
- Coenzymes reduced:



## 12.1 Entry of Pyruvate into the Mitochondrion

- **Fig 12.1** Pyruvate translocase transports pyruvate into the mitochondria in symport with  $H^+$



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## 12.2 Conversion of Pyruvate to Acetyl CoA

- Pyruvate dehydrogenase complex (PDH complex) is a multienzyme complex containing:

3 enzymes + 5 coenzymes + other proteins

(+ ATP coenzyme as a regulator)

$E_1$  = pyruvate dehydrogenase

$E_2$  = dihydrolipoamide acetyltransferase

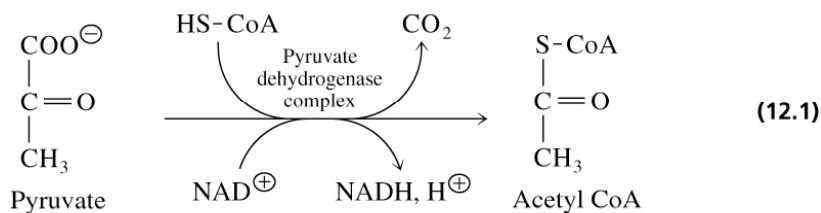
$E_3$  = dihydrolipoamide dehydrogenase

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## Overall reaction of pyruvate dehydrogenase complex



## Table 12.1 Components of the PDH Complex

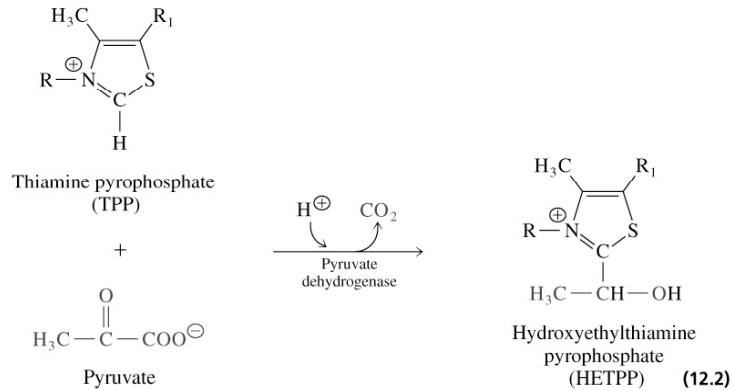
**TABLE 12.1** Components of the pyruvate dehydrogenase complex in mammals and in *E. coli*

Enzyme	Coenzyme	Oligomeric form		Number of oligomers per complex	
		Mammals	<i>E. coli</i>	Mammals	<i>E. coli</i>
Pyruvate dehydrogenase(E1)	TPP	$\alpha_2\beta_2$	$\alpha_2$	20–30	12
Dihydrolipoamide acetyltransferase(E2)	Lipoic acid, HS-CoA	$\alpha_{60}$	$\alpha_{24}$	1	1
Dihydrolipoamide dehydrogenase(E3)	FAD, $\text{NAD}^{\oplus}$	$\alpha_2$	$\alpha_2$	6	6
E3-binding protein	Lipoic acid	Monomer	—	12	—
Kinase	ATP	$\alpha\beta$	—	2–3	—
Phosphatase	None	—	—	>3	—

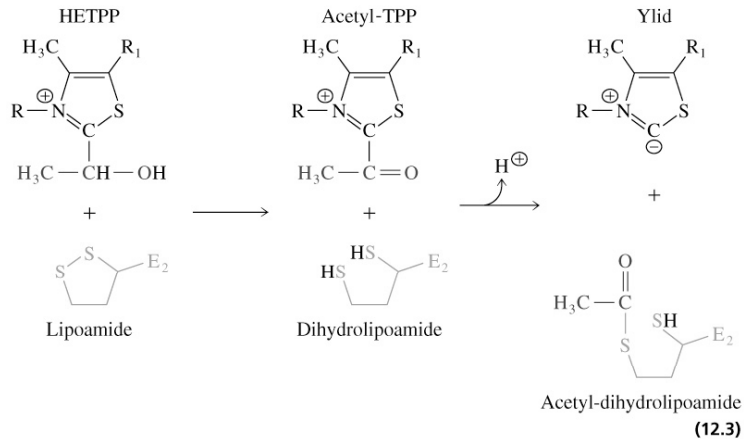
[Adapted from Patel, M. S., and Roche, T. E. (1990). Molecular biology and biochemistry of pyruvate dehydrogenase complexes. *FASEB J.* 4:3224–3233.]

# The five steps of the PDH complex

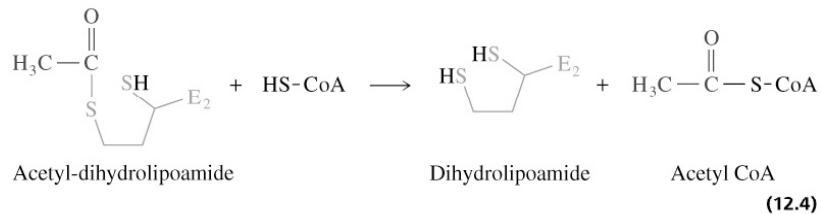
## Step 1: Catalyzed by E<sub>1</sub>



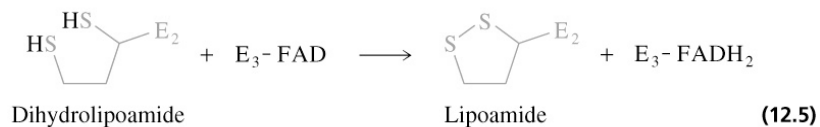
## Step 2: The second step is also catalyzed by E<sub>1</sub>



### Step 3: E<sub>2</sub> transfers the lipoamide-bound acetyl group to HS-CoA forming acetyl CoA

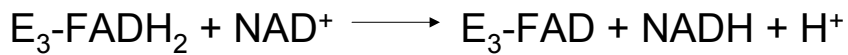


### Step 4: E<sub>3</sub> FAD group oxidizes reduced lipoamide of E<sub>2</sub> forming FADH<sub>2</sub>

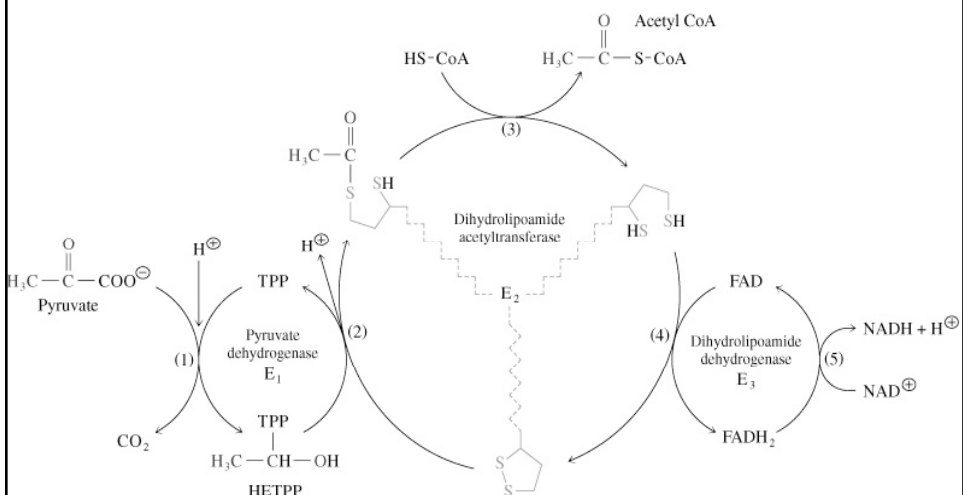


## Step 5: E<sub>3</sub>-FADH<sub>2</sub> reduces NAD<sup>+</sup> to regenerate E<sub>3</sub>-FAD and NADH

- The oxidation of E<sub>3</sub>-FADH<sub>2</sub> regenerates the original holoenzyme completing the catalytic cycle
- NADH dissociates from the complex



### Fig 12.2 Reactions of the PDH complex

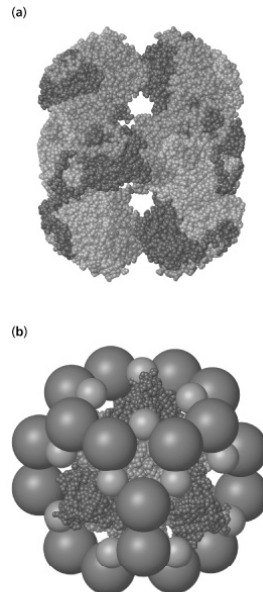


## Roles of the coenzymes of the PDH complex

- $\text{NAD}^+$  and HS-CoA are cosubstrates
- TPP, lipoamide and FAD are prosthetic groups
- ATP is a regulator of the PDH complex
- Lipoamide (on  $\text{E}_2$ ) acts as a “swinging arm” to transfer the two carbon unit from the active site of  $\text{E}_1$  to the active site of  $\text{E}_3$  (substrate channeling)

### Fig 12.3

- Structure of the PDH complex
- (a) Core of the complex (24  $\text{E}_2$  chains)
- (b) Model of the entire complex: 12  $\text{E}_1$  dimers (blue), 6  $\text{E}_3$  dimers (green) surround the core



## 12.3 The Citric Acid Cycle Oxidizes AcetylCoA

### • Table 12.2

**TABLE 12.2** The enzymatic reactions of the citric acid cycle

Reaction	Enzyme
1. Acetyl CoA + Oxaloacetate + H <sub>2</sub> O → Citrate + HS-CoA + H <sup>⊕</sup>	Citrate synthase
2. Citrate ⇌ Isocitrate	Aconitase (Aconitate hydratase)
3. Isocitrate + NAD <sup>⊕</sup> → α-Ketoglutarate + NADH + CO <sub>2</sub>	Isocitrate dehydrogenase
4. α-Ketoglutarate + HS-CoA + NAD <sup>⊕</sup> → Succinyl CoA + N + ADH CO <sub>2</sub>	α-Ketoglutarate dehydrogenase complex
5. Succinyl CoA + GDP (or ADP) + P <sub>i</sub> ⇌ Succinate + GTP (or ATP) + HS-CoA	Succinyl-CoA synthetase
6. Succinate + Q ⇌ Fumarate + QH <sub>2</sub>	Succinate dehydrogenase complex
7. Fumarate + 2H <sub>2</sub> O ⇌ L-Malate	Fumarase (Fumarate hydratase)
8. L-Malate + NAD <sup>⊕</sup> ⇌ Oxaloacetate + NADH + H <sup>⊕</sup>	Malate dehydrogenase
Net equation:	
Acetyl CoA + 3 NAD <sup>⊕</sup> + Q + GDP (or ADP) + P <sub>i</sub> + 2 H <sub>2</sub> O → HS-CoA + 3 NADH + QH <sub>2</sub> + GTP (or ATP) + 2 CO <sub>2</sub> + 2 H <sup>⊕</sup>	

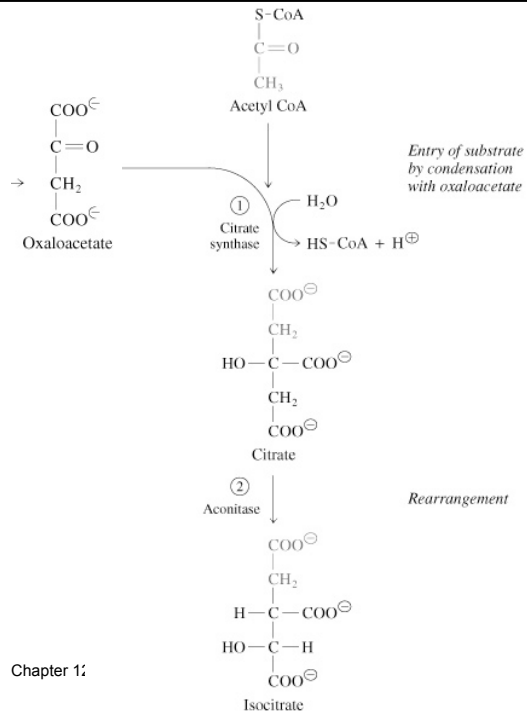
### Summary of the citric acid cycle

- For each acetyl CoA which enters the cycle:
  - (1) Two molecules of CO<sub>2</sub> are released
  - (2) Coenzymes NAD<sup>+</sup> and Q are reduced
  - (3) One GDP (or ADP) is phosphorylated
  - (4) The initial acceptor molecule (oxaloacetate) is reformed



**Fig 12.4**

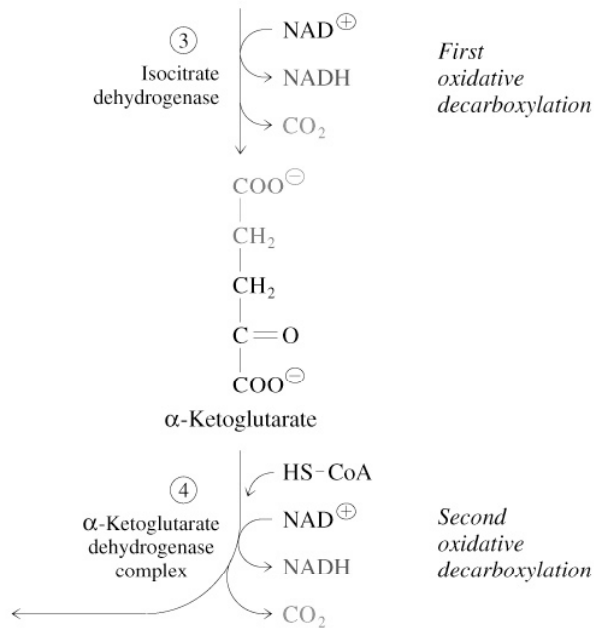
- Citric acid cycle (Four slides)



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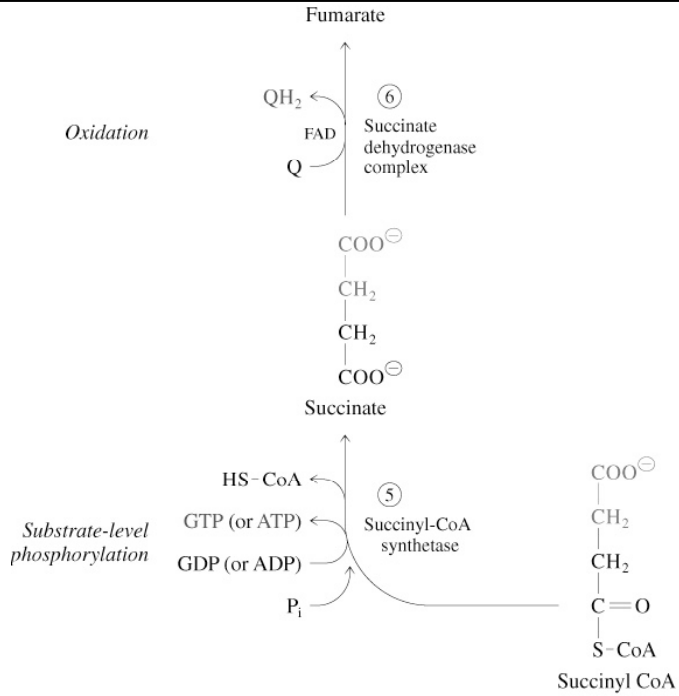
Chapter 1:

**Fig 12.4 (continued)**

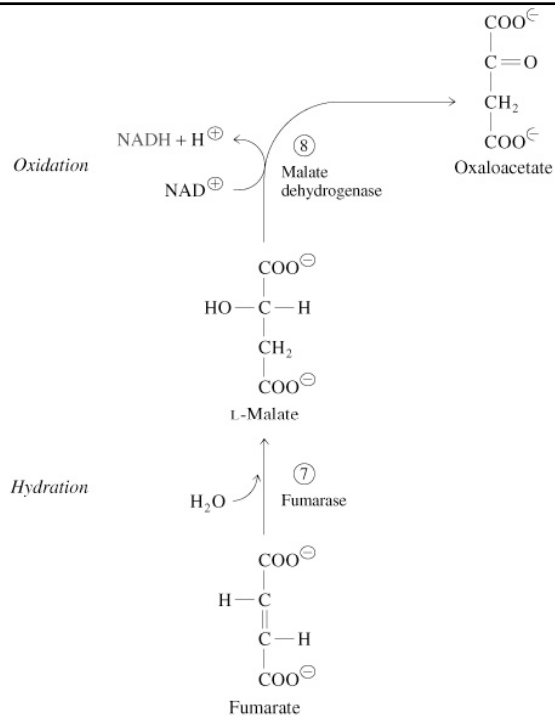


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**Fig 12.4  
(continued)**

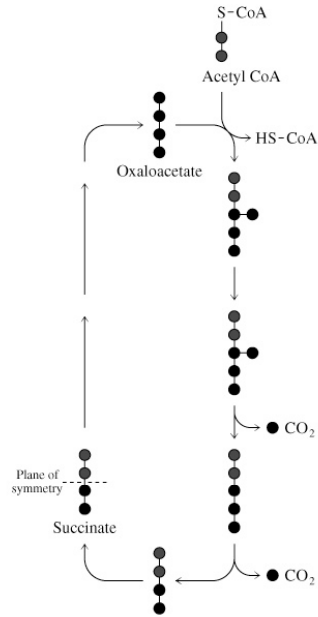


**Fig 12.4  
(continued)**



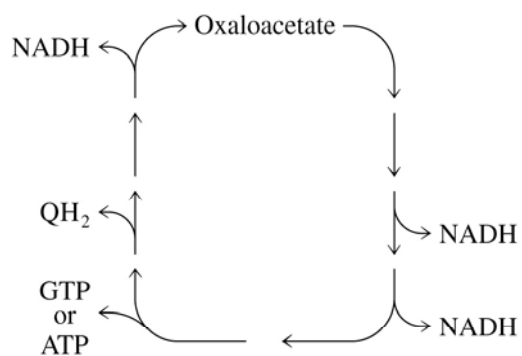
**Fig 12.5**

- Fates of carbon atoms in the cycle
- Carbon atoms from acetyl CoA (red) are not lost in the first turn of the cycle



**Fig 12.6 Energy conservation by the cycle**

- Energy is conserved in the reduced coenzymes NADH, QH<sub>2</sub> and one GTP
- NADH, QH<sub>2</sub> can be oxidized to produce ATP by oxidative phosphorylation

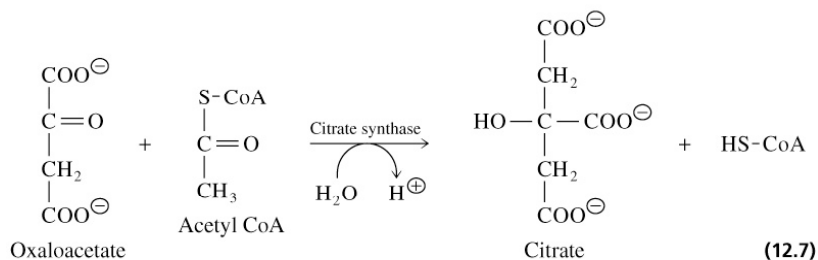


## 12.4 The Citric Acid Cycle Can Be a Multistep Catalyst

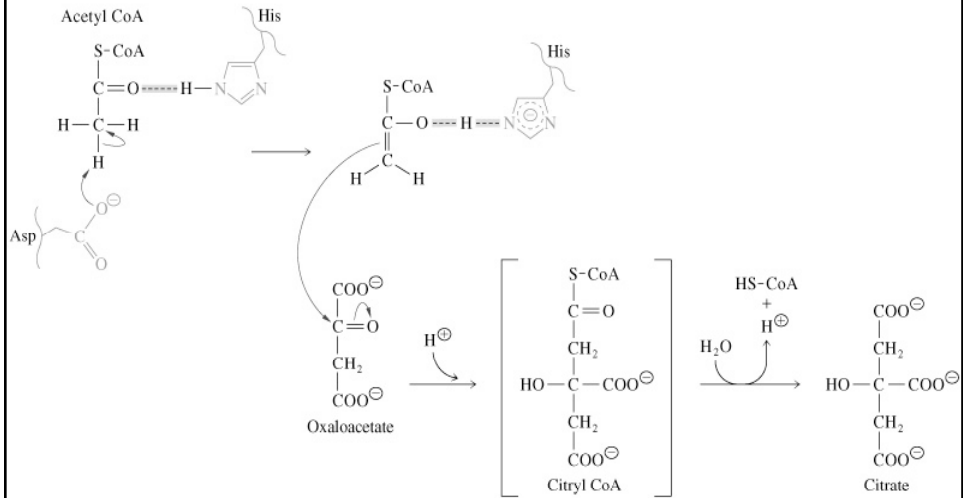
- Oxaloacetate is regenerated
- The cycle is a mechanism for oxidizing acetyl CoA to CO<sub>2</sub> by NAD<sup>+</sup> and Q
- The cycle itself is not a pathway for a net degradation of any cycle intermediates
- Cycle intermediates can be shared with other pathways, which may lead to a resupply or net decrease in cycle intermediates

### 1. Citrate Synthase

- Citrate formed from acetyl CoA and oxaloacetate
- Only cycle reaction with C-C bond formation



**Fig 12.7 Proposed mechanism of citrate synthase**



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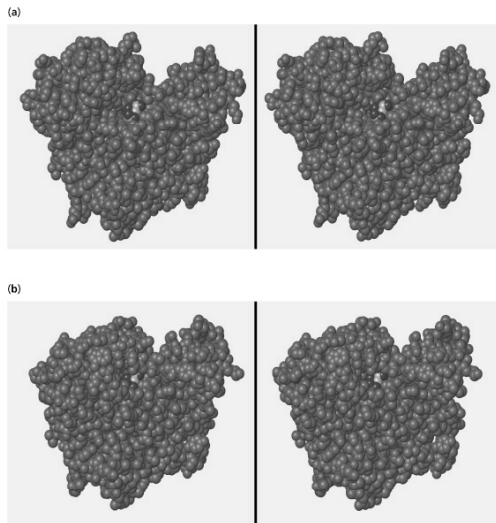
25

**Fig 12.8 Stereo views of citrate synthase**

(a) Open conformation

(b) Closed conformation

Product citrate (red)



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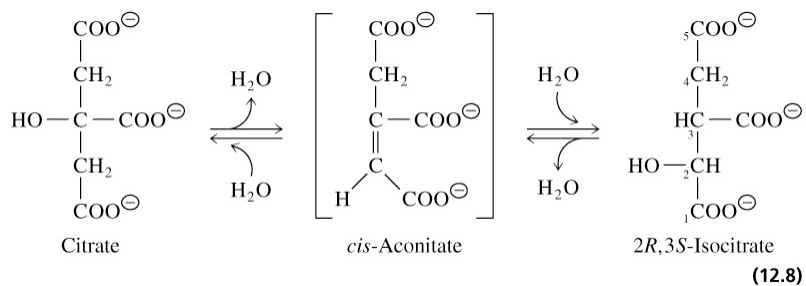
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## 2. Aconitase

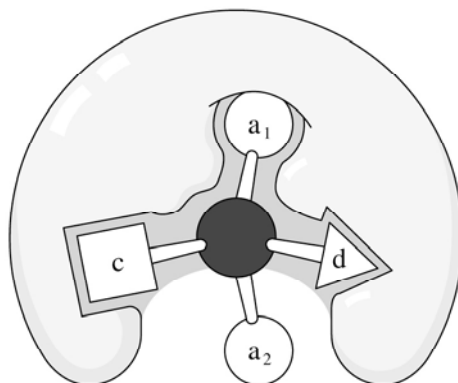
- Elimination of H<sub>2</sub>O from citrate to form C=C bond of *cis*-aconitate
- Stereospecific addition of H<sub>2</sub>O to *cis*-aconitate to form 2*R*,3*S*-Isocitrate

## Reaction of Aconitase



## Box 12.1 Three point attachment of prochiral substrates to enzymes

- Chemically identical groups  $a_1$  and  $a_2$  of a prochiral molecule can be distinguished by the enzyme

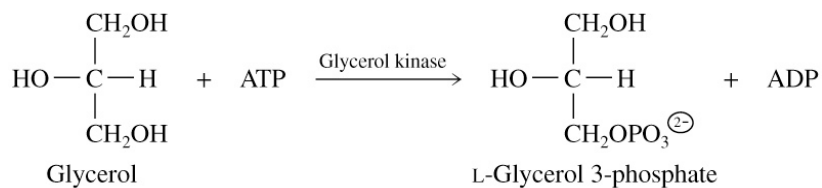


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## Glycerol kinase converts prochiral substrate (glycerol) into a chiral product (L-glycerol 3-phosphate)



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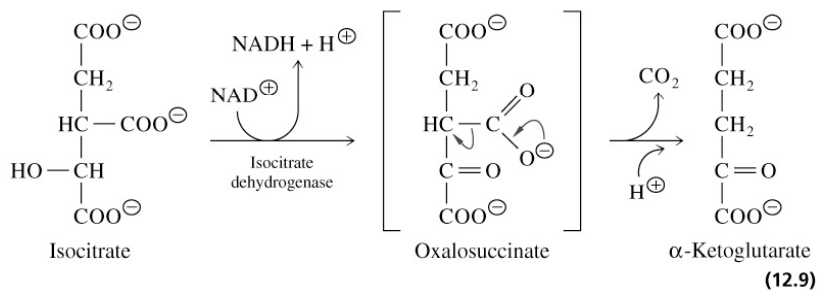
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### 3. Isocitrate Dehydrogenase

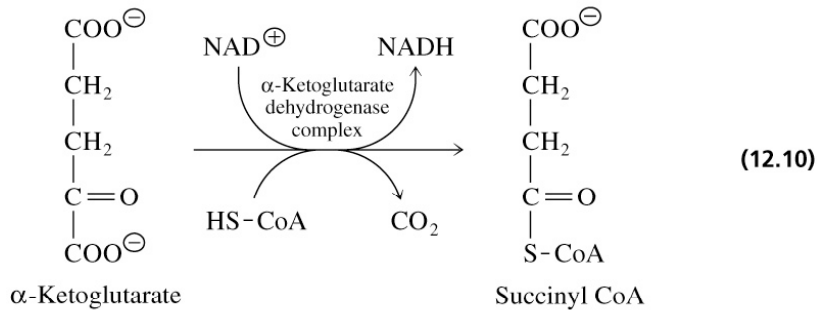
- Oxidative decarboxylation of isocitrate to  $\alpha$ -ketoglutarate ( $\alpha$ -kg) (a metabolically irreversible reaction)
- One of four oxidation-reduction reactions of the cycle
- Hydride ion from the C-2 of isocitrate is transferred to  $\text{NAD}^+$  to form NADH
- Oxalosuccinate is decarboxylated to  $\alpha$ -kg

### Isocitrate dehydrogenase reaction





## 4. The $\alpha$ -Ketoglutarate Dehydrogenase Complex



### Structure of $\alpha$ -Ketoglutarate dehydrogenase complex

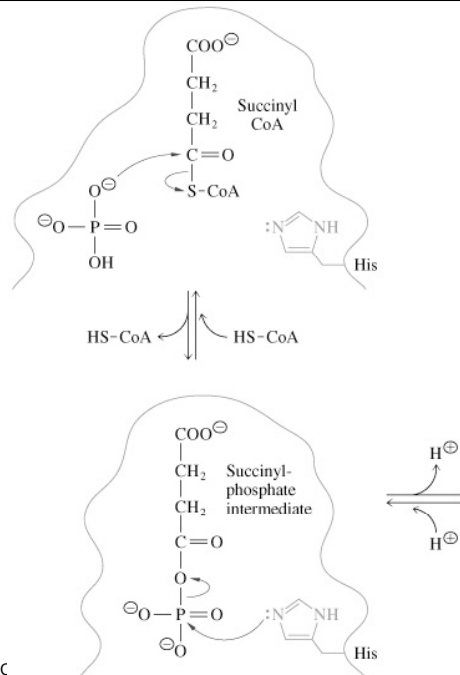
- Similar to pyruvate dehydrogenase complex
  - Same coenzymes, identical mechanisms
- $E_1$  -  $\alpha$ -ketoglutarate dehydrogenase (with TPP)
- $E_2$  - succinyltransferase (with flexible lipoamide prosthetic group)
- $E_3$  - dihydrolipoamide dehydrogenase (with FAD)

## 5. Succinyl-CoA Synthetase

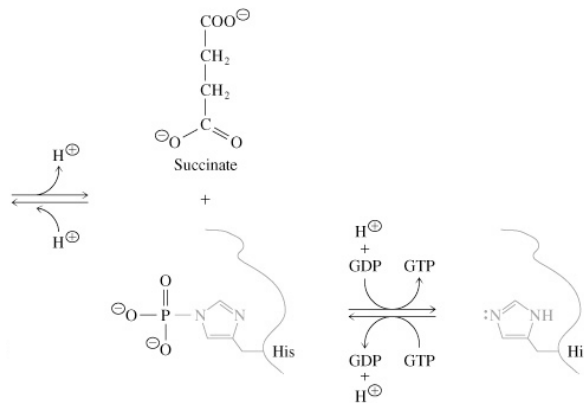
- Free energy in thioester bond of succinyl CoA is conserved as GTP (or ATP in plants, some bacteria)

**Fig 12.9**

- Mechanism of succinyl-CoA synthetase (continued on next slide)



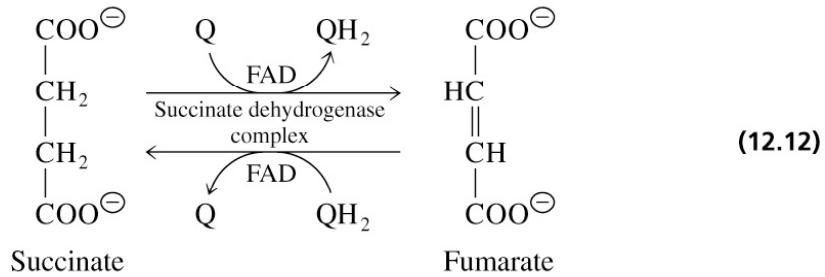
**Fig 12.9 (continued)**



## 6. The Succinate Dehydrogenase (SDH) Complex

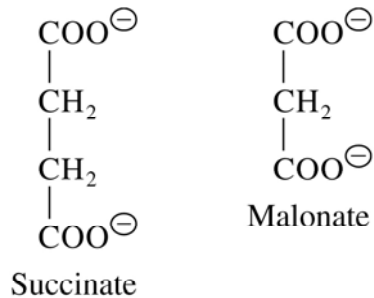
- Located on the inner mitochondrial membrane (other components are dissolved in the matrix)
- Dehydrogenation is stereospecific; only the *trans* isomer is formed
- Substrate analog malonate is a competitive inhibitor of the SDH complex

## Reaction of the succinate dehydrogenase complex



## Fig 12.10 Succinate and malonate

- Malonate is a structural analog of succinate
- Malonate binds to the enzyme active site, and is a competitive inhibitor

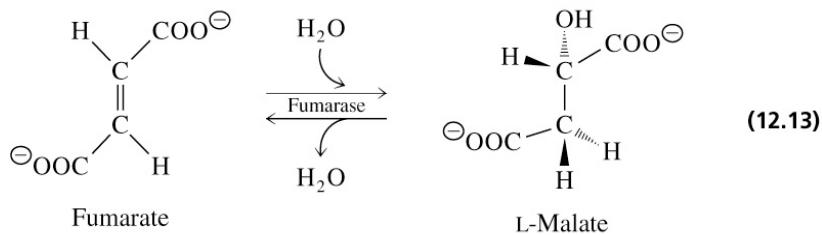


## Structure of the SDH complex

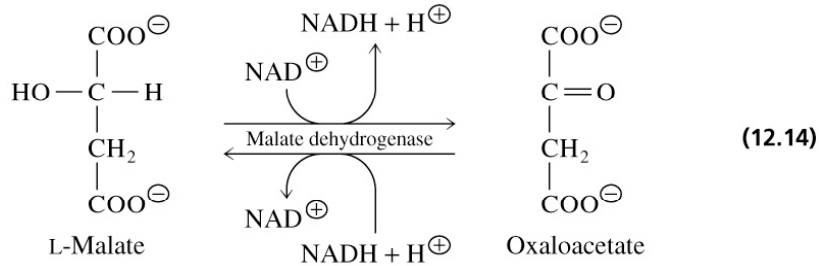
- Complex of several polypeptides, an FAD prosthetic group and iron-sulfur clusters
- Electrons are transferred from succinate to ubiquinone (Q), a lipid-soluble mobile carrier of reducing power
- FADH<sub>2</sub> generated is reoxidized by Q
- QH<sub>2</sub> is released as a mobile product

## 7. Fumarase

- Stereospecific *trans* addition of water to the double bond of fumarate to form *L*-malate



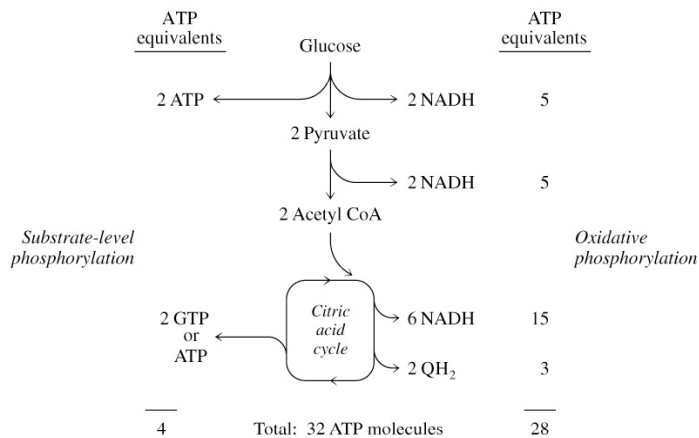
## 8. Malate Dehydrogenase



## 12.5 Reduced Coenzymes Fuel the Production of ATP

- Each acetyl CoA entering the cycle nets:
  - (1) **3 NADH**
  - (2) **1 QH<sub>2</sub>**
  - (3) **1 GTP (or 1 ATP)**
- Oxidation of each NADH yields **2.5 ATP**
- Oxidation of each QH<sub>2</sub> yields **1.5 ATP**
- Complete oxidation of **1 acetyl CoA = 10 ATP**

**Fig 12.11 Glucose degradation via glycolysis, citric acid cycle, and oxidative phosphorylation**



## **NADH fate in anaerobic glycolysis**

### Anaerobic glycolysis

- NADH produced by G3PDH reaction is reoxidized to NAD<sup>+</sup> in the pyruvate to lactate reaction
- NAD<sup>+</sup> recycling allows G3PDH reaction (and glycolysis) to continue anaerobically

## NADH fate in aerobic glycolysis

- Glycolytic NADH is not reoxidized via pyruvate reduction but is available to fuel ATP formation
- Glycolytic NADH (cytosol) must be transferred to mitochondria (electron transport chain location)
- Two NADH shuttles are available (next slide)

## NADH shuttles

- Malate-aspartate shuttle (most common)

**One cytosolic NADH yields ~ 2.5 ATP**  
(total **32 ATP/glucose**)

- Glycerol phosphate shuttle

**One cytosolic NADH yields ~1.5ATP**  
(total **30 ATP/glucose**)

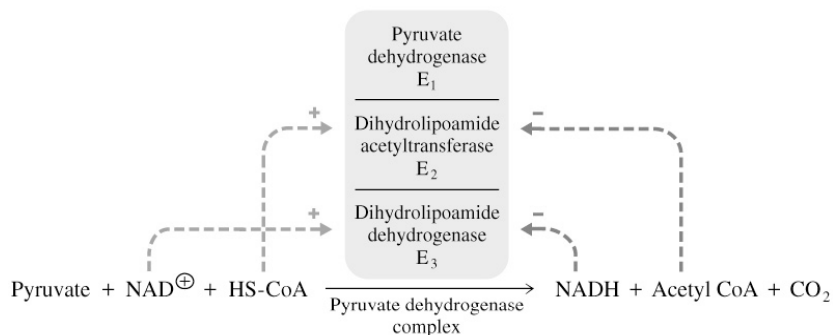


## 12.6 Regulation of the Citric Acid Cycle

- Pathway controlled by:
  - (1) Allosteric modulators
  - (2) Covalent modification of cycle enzymes
  - (3) Supply of acetyl CoA
  - (4) Regulation of pyruvate dehydrogenase complex controls acetyl CoA supply

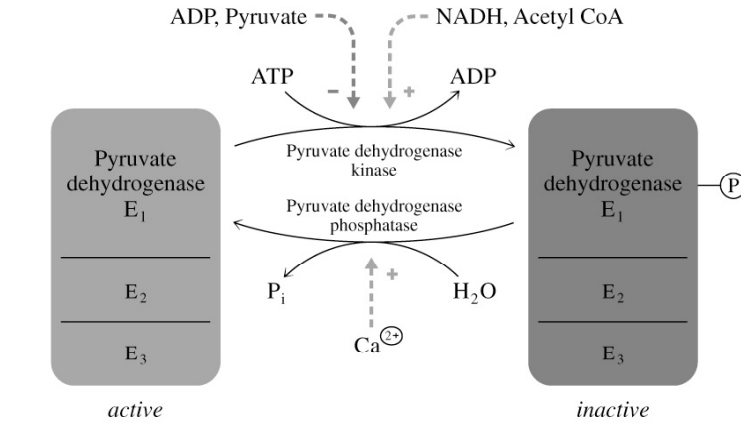
### Fig 12.12 Regulation of the PDH complex

- Increased levels of acetyl CoA and NADH inhibit  $E_2$ ,  $E_3$  in mammals and *E. coli*



## Fig 12.13 Regulation of mammalian PDH complex by covalent modification

- Phosphorylation/dephosphorylation of E<sub>1</sub>



## Further regulation of the PDH complex

### Pyruvate dehydrogenase kinase (PDK)

- PDK is activated by NADH and acetyl CoA (leads to inactivation of the PDH complex)
- PDK is inhibited by pyruvate and ADP (leads to activation of the PDH complex)

### Pyruvate dehydrogenase phosphatase (PDP)

- PDP activity is stimulated by Ca<sup>2+</sup> (leads to an activation of the PDH complex)

## Regulation of isocitrate dehydrogenase (ICDH)

### Mammalian ICDH

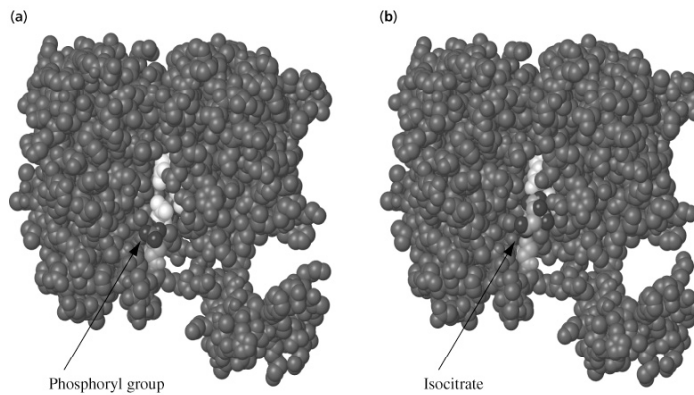
- Allosteric effectors: (+)  $\text{Ca}^{2+}$ , ADP, (-) NADH  
(not subject to covalent modification)

### *E. coli* ICDH

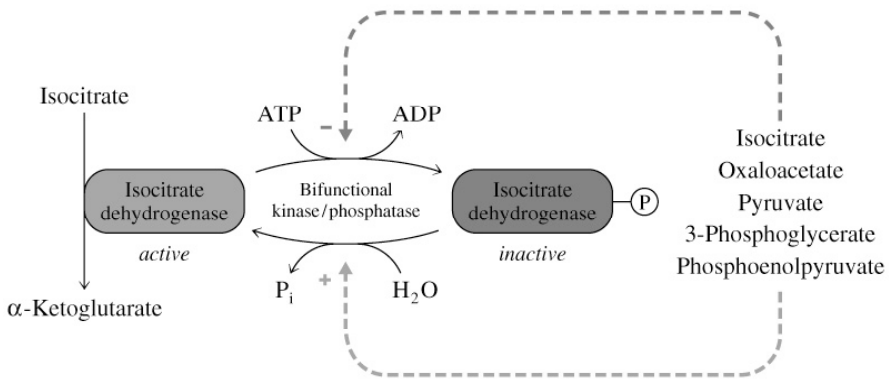
- Bifunctional kinase/phosphatase enzyme phosphorylates/dephosphorylates ICDH
- Bifunctional enzyme is reciprocally regulated by intermediates of glycolytic and citric acid cycles

## Fig 12.14 *E. coli* ICDH models

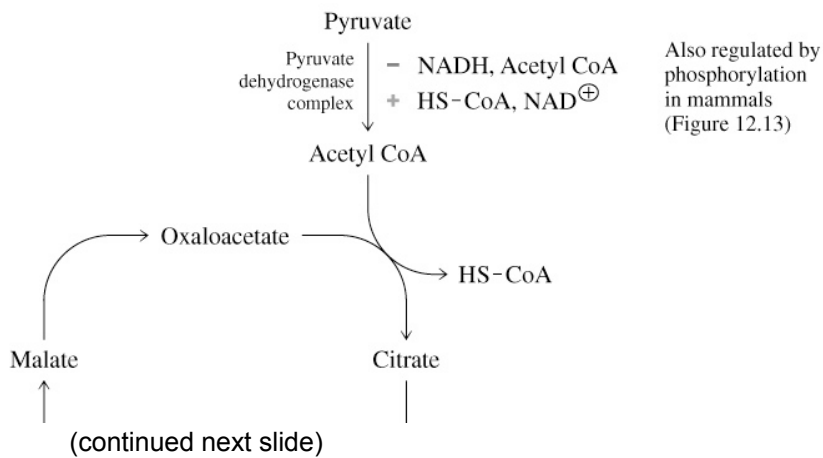
(a) Phosphorylated form inactive, (b) Active form



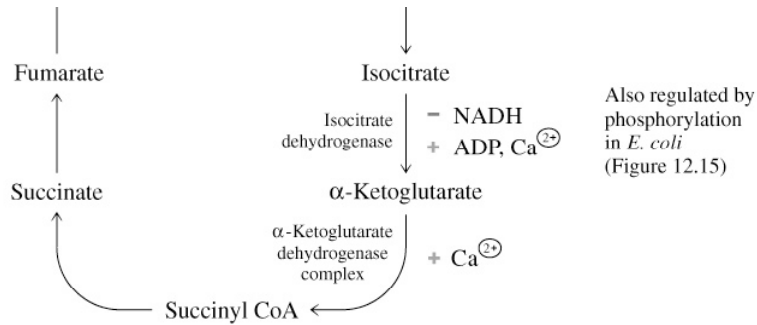
**Fig 12.15 Covalent regulation of *E. coli* ICDH**



**Fig 12.16 Regulation of the PDH complex and the citric acid cycle**



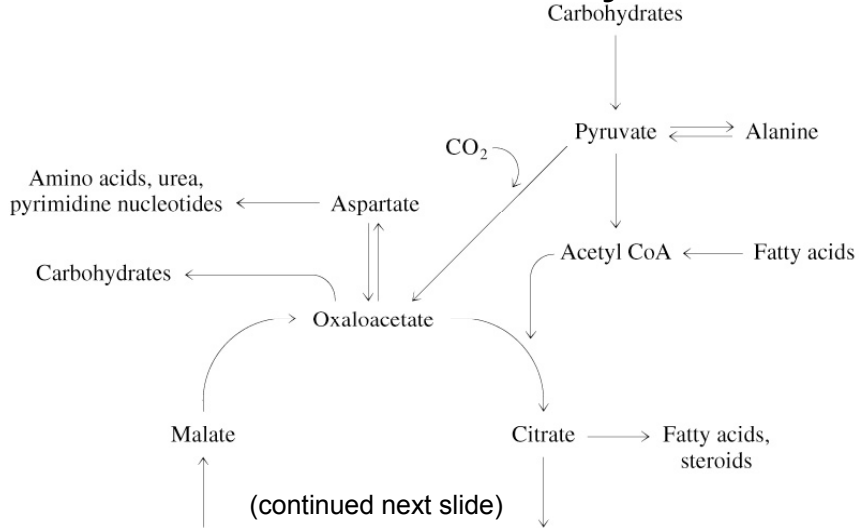
**Fig 12.16 (continued)**



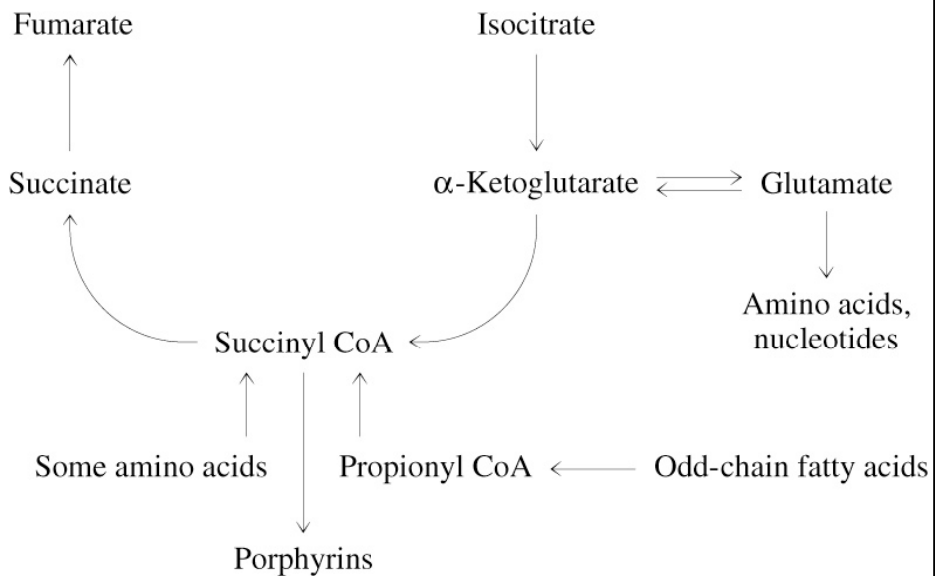
## 12.7 Entry and Exit of Metabolites

- Intermediates of the citric acid cycle are precursors for carbohydrates, lipids, amino acids, nucleotides and porphyrins
- Reactions feeding into the cycle replenish the pool of cycle intermediates

**Fig 12.17 Routes leading to and from the citric acid cycle**

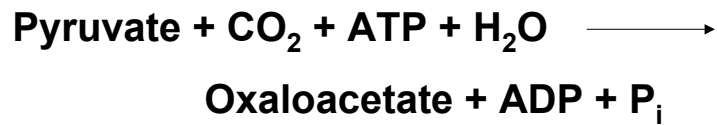


**Fig. 12.17 (continued)**



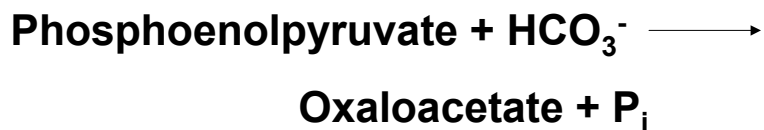
## Anaplerotic reactions

- **Anaplerotic** (filling up) reactions replenish citric acid cycle intermediates
- **Pyruvate carboxylase** is a major anaplerotic reaction in mammalian tissues



## Phosphoenolpyruvate carboxylase

- Anaplerotic reaction in plants and bacteria
- Supplies oxaloacetate to the citric acid cycle



## 12.8 The Glyoxylate Cycle

- Pathway for the formation of glucose from noncarbohydrate precursors in plants, bacteria and yeast (not animals)
- Glyoxylate cycle leads from 2-carbon compounds to glucose
- In animals, acetyl CoA is not a carbon source for the net formation of glucose (2 carbons of acetyl CoA enter cycle, 2 are released as 2 CO<sub>2</sub>)

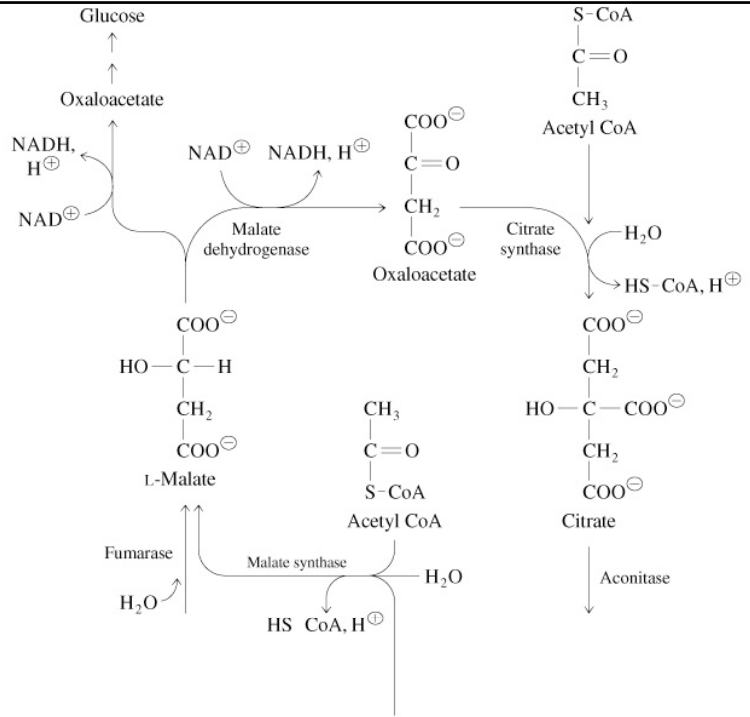
## Glyoxylate cycle - formation of glucose

- Formation of glucose from acetyl CoA (or any substrate that is a precursor to acetyl CoA)
- Ethanol or acetate can be metabolized to acetyl CoA and then to glucose via the glyoxylate cycle
- Stored seed oils in plants are converted to carbohydrates during germination



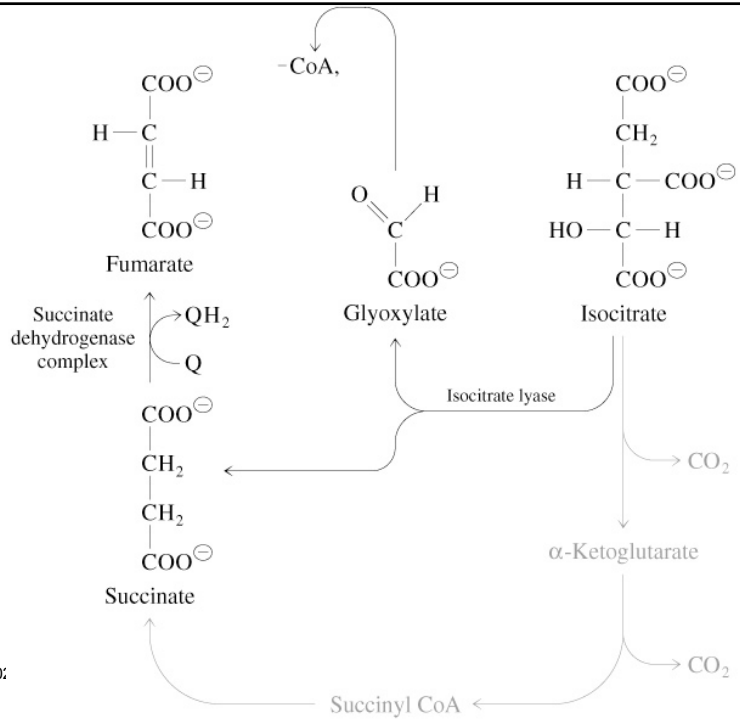
**Fig 12.18**

**Glyoxylate Cycle  
(2 slides)**



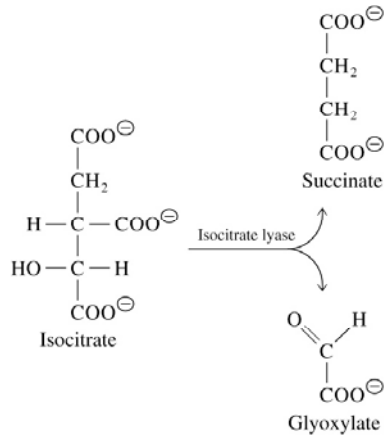
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**Fig 12.18  
(cont)**

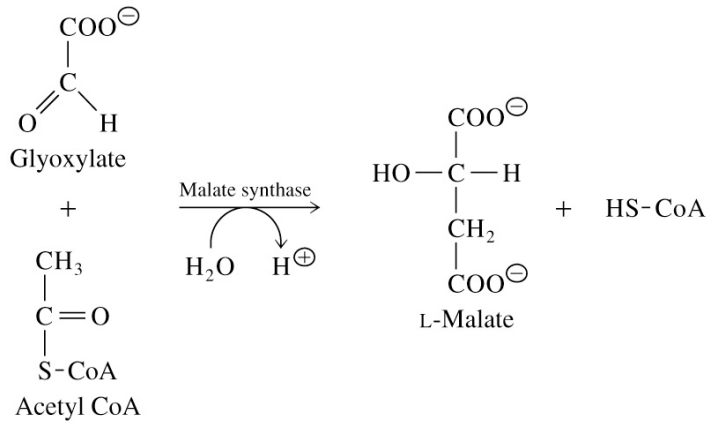


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## Fig 12.19 Isocitrate lyase: first bypass enzyme of glyoxylate



## Fig 12.20 Malate synthase: second bypass enzyme of glyoxylate



## Glyoxylate cycle in germinating castor beans

- Figure 12.21 (next slide)
- Conversion of acetyl CoA to glucose requires the transfer of metabolites among three metabolic compartments
  - (1) The glyoxysome
  - (2) The cytosol
  - (3) The mitochondrion

